

ON A COMPUTATIONAL METHOD OF PAIRS OF EIGENVECTORS ASSOCIATED WITH A DOUBLE DEGENERATE EIGENVALUE OF DIAGONIZABLE SQUARE MATRIX. APPLICATION TO HUCKEL'S MATRIX OF ALTERNANT HYDROCARBONS.

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I. DEFINITIONS.

We consider a square matrix  $M = (m_{ik})$ . Its eigenvalues are  $y_\alpha, y_\beta, \dots, y_\mu, \dots$ . The eigenvectors associated to these eigenvalues are  $X_\alpha, X_\beta, \dots, X_\mu, \dots$ . The components of the eigenvector  $X_\mu$  are  $x_{1\mu}, x_{2\mu}, \dots, x_{i\mu}, \dots$

$M^t$ , transpose of  $M$  have the same eigenvalues. Its eigenvectors are  $Z_\alpha, Z_\beta, \dots, Z_\mu, \dots$ . The components of  $Z_\mu$  are  $z_{1\mu}, z_{2\mu}, \dots, z_{i\mu}, \dots$

II. REMINDER.

$F$ , function of variable  $y$  and matrix elements  $m_{ik}$  is the characteristic polynomial of  $M$  and  $M^t$ .

$y_\mu$  being a non degenerate eigenvalue, we have

$$(1) \quad x_{k\mu} z_{i\mu} = - \left[ \frac{\frac{\partial F}{\partial m_{ik}}}{\frac{\partial F}{\partial y}} \right]_{y=y_\mu}$$

If  $y_\mu$  is a double degenerate eigenvalue, the eigenvectors associated to  $y$  are resp.  $X_{\mu'}, X_{\mu''}$ , and  $Z_{\mu'}, Z_{\mu''}$  and we have the relation

$$(2) \quad P_\mu = \left[ m_{ki} \right]_{y=y_\mu} = \left[ x_{k\mu'} z_{i\mu'} + x_{k\mu''} z_{i\mu''} \right]_{y=y_\mu} = -2 \left[ \frac{\frac{\partial^2 F}{\partial y \partial m_{ik}}}{\frac{\partial^2 F}{\partial y^2}} \right]_{y=y_\mu}$$

The double degenerate  $y_\mu$  being root of  $\frac{\partial F}{\partial y}$  and  $\frac{\partial F}{\partial m_{ik}}$  we consider  $\frac{\partial F}{\partial y} = (y - y_\mu) \Phi$  and we have

$$(3) \quad \frac{\partial^2 F}{\partial y^2} = \Phi + (y - y_\mu) \frac{d\Phi}{dy}$$

$$(4) \quad \left[ \frac{\partial^2 F}{\partial y^2} \right]_{y=y_\mu} = \Phi$$

$$(5) \quad (y - y_\mu) \left[ \frac{\partial^2 F}{\partial y^2} \right]_{y=y_\mu} = \left[ \frac{\partial F}{\partial y} \right]_{y=y_\mu}$$

In the same way we show

$$(6) \frac{\partial^2 F}{\partial y \partial m_{ik}} = (y - y_{\mu}) \left[ \frac{\partial F}{\partial m_{ik}} \right]_{y=y_{\mu}}$$

Considering (5) and (6), relation (2) becomes

$$(7) P_{\mu} = [r_{\mu k}]_{y=y_{\mu}} = [\alpha_{\mu k} z_{i\mu}' + \alpha_{\mu k} \beta_{i\mu}']_{y=y_{\mu}} = -2 \left[ \frac{\partial F}{\partial m_{ik}} \frac{\partial F}{\partial y} \right]_{y=y_{\mu}}$$

III. DETERMINATION OF PAIRS OF EIGENVECTORS ASSOCIATED WITH THE DEGENERATE EIGENVALUES.

a) Principle. We explain a computational method of determination of eigenvectors  $X_{\mu}'$  and  $Z_{\mu}'$ ; we obtain the matrix

$$(8) P_{\mu}' = [\alpha_{\mu k} z_{i\mu}']_{y=y_{\mu}} = X_{\mu}' Z_{\mu}'$$

The matrix  $P_{\mu}'' = P_{\mu} - P_{\mu}'$  determines the eigenvectors  $X_{\mu}''$  and  $Z_{\mu}''$ .

b) Determination of eigenvectors  $X_{\mu}'$  and  $Z_{\mu}'$ .

Substituting in M the diagonal element  $m_{nn}$  by  $m_{nn} + q_n$ , we obtain matrix N. F being the characteristic polynomial of M, this polynomial for N is  $F + q_n \frac{\partial F}{\partial m_{nn}}$

$y_{\mu}$  being a doubly degenerate eigenvalue of M, we have

$$(9) (F)_{y=y_{\mu}} = \left( \frac{\partial F}{\partial m_{nn}} \right)_{y=y_{\mu}} = \left( \frac{\partial F}{\partial y} \right)_{y=y_{\mu}} = 0$$

$y_{\mu}$  is an eigenvalue of N.

The relation between eigenvectors  $U_{\mu}$  of N and  $V_{\mu}$  of  $N^t$ , transpose of N, associated to  $y_{\mu}$  is

$$(10) U_{\mu k} V_{i\mu} = - \left[ \frac{\frac{\partial F}{\partial m_{ik}} + q_n \frac{\partial^2 F}{\partial m_{ik} \partial m_{nn}}}{\frac{\partial F}{\partial y} + q_n \frac{\partial^2 F}{\partial y \partial m_{nn}}} \right]_{y=y_{\mu}}$$

$y_{\mu}$  being a doubly degenerate eigenvalue of N,  $\frac{\partial F}{\partial m_{ik}}$  being equal to zero

$$(11) U_{\mu k} V_{i\mu} = - \left[ \frac{\frac{\partial^2 F}{\partial m_{ik} \partial m_{nn}}}{\frac{\partial^2 F}{\partial y \partial m_{nn}}} \right]_{y=y_{\mu}}$$

$U_{\mu}$  and  $V_{\mu}$  being independent of  $q_n$ , are eigenvectors of M resp.  $M^t$ , associated with eigenvalue  $y_{\mu}$ .

$$(12) U_{\mu} = X_{\mu}'; V_{\mu} = Z_{\mu}'$$

Comparing relation (11) to relation (1), we see that

$$M = \begin{bmatrix} 3 & 2 & 2 & -4 \\ 2 & 3 & 2 & -1 \\ 1 & 1 & 2 & -1 \\ 2 & 2 & 2 & -1 \end{bmatrix}$$

$$X_{\beta} = \begin{bmatrix} -2 \\ 4 \\ 1 \\ 2 \end{bmatrix}$$

$$X_{\gamma} = \begin{bmatrix} 0 \\ 3 \\ 1 \\ 2 \end{bmatrix}$$

$\lambda_{\alpha} = 1 = \text{double degenerate eigenvalue}$

$$Z_{\beta} = \begin{bmatrix} -1 \\ -1 \\ -1 \\ 2 \end{bmatrix}$$

$$Z_{\gamma} = \begin{bmatrix} 1 \\ 1 \\ 1 \\ -3/2 \end{bmatrix}$$

$\lambda_{\beta} = 2$   
 $\lambda_{\gamma} = 3$

Table 1

the characteristic polynomial of  $N$  is  $\dots$ . We can choose  $N$  as  
 the matrix obtained by annulation  $\dots$  in  $M$  of all  
 elements in line and column index  $r$ .

IV. NUMERICAL EXAMPLE. We consider the matrix  $M$  table 1.

a) We have  $\left[ \frac{\partial^2 F}{\partial y \partial m_{1,1}} \right]_{y=1} = [-3y^2 + 8y - 3]_{y=1} = 2$

(13)  $\left[ \frac{\partial^2 F}{\partial y \partial m_{2,2}} \right]_{y=1} = [-3y^2 + 8y - 9]_{y=1} = -4$   $\left[ \frac{\partial^2 F}{\partial y \partial m_{3,3}} \right]_{y=1} = [-3y^2 + 10y - 9]_{y=1} = -2$

$\left[ \frac{\partial^2 F}{\partial y \partial m_{4,4}} \right]_{y=1} = [-3y^2 + 16y - 13]_{y=1} = 0$

Our method can be applied to indices 1, 2 and 3, but not to index 4, relation (11) not being applicable if  $\frac{\partial^2 F}{\partial y \partial m_{4,4}} = 0$

b) Operation index 3

(14)  $P_{\alpha'} = \frac{\partial^2 F}{\partial m_{1,1} \partial m_{3,3}} = \begin{bmatrix} -1 & -2 & 0 & 3 \\ 1 & 2 & 0 & 3 \\ 0 & 0 & 0 & 0 \\ 0 & 0 & 0 & 0 \end{bmatrix}$

columns are  $kX_{\alpha'}$ , lines are  $kZ_{\alpha'}$ .

We chose  $X_{\alpha'}$  on  $Z_{\alpha'}$  in manner that  $X_{\alpha'} Z_{\alpha'} = Z_{\alpha'} X_{\alpha'} = 1$

We have

$$(15) \quad X_{\alpha'} = \begin{vmatrix} 1 \\ -1 \\ 0 \\ 0 \end{vmatrix} \quad Z_{\alpha'} = \begin{vmatrix} -1 \\ -2 \\ 0 \\ 3 \end{vmatrix}$$

Applying relation (2) to  $y_{\alpha} = 1$ , we obtain

$$(16) \quad P_{\alpha} = \begin{vmatrix} -1 & -2 & -2 & 2 \\ 1 & 2 & 1 & -\frac{7}{2} \\ 0 & 0 & 1 & -\frac{1}{2} \\ 0 & 0 & 0 & 0 \end{vmatrix}$$

$$(17) \quad P_{\alpha''} = P_{\alpha} - P_{\alpha'} = X_{\alpha''} Z_{\alpha''}' = \begin{vmatrix} 0 & 0 & -2 & 1 \\ 0 & 0 & 1 & -\frac{1}{2} \\ 0 & 0 & 1 & -\frac{1}{2} \\ 0 & 0 & 0 & 0 \end{vmatrix}$$

Choosing  $X_{\alpha''} = \begin{vmatrix} 2 \\ -1 \\ -1 \\ 0 \end{vmatrix}$ , we have  $Z_{\alpha''}' = \begin{vmatrix} 0 \\ 0 \\ -\frac{1}{2} \\ \frac{1}{2} \end{vmatrix}$

Sets of eigenvectors of M are given in table 2.

$X_{\alpha'} = \begin{vmatrix} 1 \\ -1 \\ 0 \\ 0 \end{vmatrix}$	$X_{\alpha''} = \begin{vmatrix} 2 \\ -1 \\ -1 \\ 0 \end{vmatrix}$	$X_{\beta} = \begin{vmatrix} -2 \\ 4 \\ 1 \\ 2 \end{vmatrix}$	$X_{\gamma} = \begin{vmatrix} 0 \\ 3 \\ 1 \\ 2 \end{vmatrix}$
$Z_{\alpha'} = \begin{vmatrix} -1 \\ -2 \\ 0 \\ 3 \end{vmatrix}$	$Z_{\alpha''}' = \begin{vmatrix} 0 \\ 0 \\ -\frac{1}{2} \\ \frac{1}{2} \end{vmatrix}$	$Z_{\beta} = \begin{vmatrix} -1 \\ -1 \\ -1 \\ 2 \end{vmatrix}$	$Z_{\gamma} = \begin{vmatrix} 1 \\ 1 \\ 1 \\ -\frac{3}{2} \end{vmatrix}$

Table 2

These sets verify the relation  $X_{\mu}' Z_{\nu} = Z_{\mu}' X_{\nu} = \delta_{\mu\nu}$   
 $\delta_{\mu\nu}$  being Kroneckers symbol.

c) Operation indices 1 and 2. Results see table 3

$X_{\alpha^i} = \begin{vmatrix} 0 \\ -1 \\ 1 \\ 0 \end{vmatrix}$	$X'_{\alpha^{ii}} = \begin{vmatrix} 1 \\ -1 \\ 0 \\ 0 \end{vmatrix}$	} Index 1
$Z_{\alpha^i} = \begin{vmatrix} 0 \\ 0 \\ 1 \\ -1/2 \end{vmatrix}$	$Z_{\alpha^{ii}} = \begin{vmatrix} -1 \\ -2 \\ -2 \\ 4 \end{vmatrix}$	
$X_{\alpha^i} = \begin{vmatrix} 1 \\ 0 \\ -1 \\ 0 \end{vmatrix}$	$X_{\alpha^{ii}} = \begin{vmatrix} 1 \\ -1 \\ 0 \\ 0 \end{vmatrix}$	} Index 2
$Z_{\alpha^i} = \begin{vmatrix} 0 \\ 0 \\ -1 \\ 1/2 \end{vmatrix}$	$Z_{\alpha^{ii}} = \begin{vmatrix} -1 \\ -2 \\ -1 \\ 7/2 \end{vmatrix}$	

Table 3

V. APPLICATION TO THE DOUBLE DEGENERATE EIGENVALUES OF SYMMETRIC MATRICES USED IN THE HUCKEL'S METHOD (2) APPLIED TO ALTERNANT HYDROCARBONS.

We shall recall the relations for symmetric matrices.

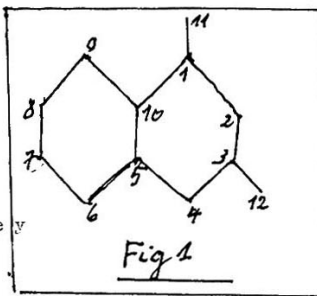
$$X_{\mu} = Z_{\mu} \text{ and } X_{\mu}^i X_{\nu} = \delta_{\mu\nu}$$

In most cases the symmetry of the molecule permits the decomposition of the matrix in several matrices corresponding to the irreducible representations of the symmetric group of the molecule. If two different representations correspond to the degenerate eigenvalue  $y_{\mu}$ , the eigenvectors  $X_{\mu}^i$  and  $X_{\mu}^j$  are eigenvectors of non-degenerate eigenvalue  $y_{\mu}$  of two different matrices. But if  $y_{\mu}$  is a doubly degenerate eigenvalue of a matrix corresponding to a same irreducible representation, our computational method (as others<sup>(4)</sup>) may be used.

Example. 1/3 Naphtoquinodimethane.

Application of Huckel's H.M.O. method leads to matrix represented by graph fig. 1. Vertex index i represents diagonal element  $m_{ii}$

of the matrix. Edge  $ij$  represents the element  $m_{ij}$  and  $m_{ji}$ . All diagonal elements of  $M$  are equal to zero. Non diagonal elements  $m_{ij} = 1$ , if  $ij$  is an edge of the graph;  $m_{ij} = 0$  in the other case. Table 4 gives eigenvalues and eigenvectors of this matrix. We consider the eigenvalue



$\gamma_p = 0$ , which is doubly degenerate.

Our method gives no results, if we operate to indices 1, 3, 5, 7 or 9 = 0

if  $r = 1, 3, 5, 7$  or 9.

Table 5 gives graphs of pairs of eigenvectors, associated to the degenerate eigenvalue  $\gamma_p = 0$

- A) Pair of eigenvectors corresponding table 4
- B) Pair obtained by operation index 2
- C) " " " " INDICES 4, 6, 8, 10
- D) " " " " Index 11
- E) " " " " index 12

$\gamma_p$	1	2	3	4	5	6	7	8
$\alpha_i$	2.37953	1.76767	1.50078	1.12413	.83343	.00000	-.00001	-.83343
1	.26858	-.19926	-.49915	-.02362	-.27534	-.00000	.00001	.27584
2	.27769	-.37111	-.11597	.24025	-.20634	-.51820	-.27263	-.20624
3	.29430	-.45691	.32510	.29668	.10386	.00000	.00001	-.10386
4	.29905	-.17822	.38725	-.17057	.16829	.35518	-.46551	.16829
5	.41740	.14185	.25608	-.48863	.03639	.00000	-.00001	-.03639
6	.25362	.29732	.29762	-.12221	-.44537	-.17759	.23276	-.44527
7	.18620	.38557	.19053	.33999	-.40758	-.00000	-.00001	.40758
8	.18950	.38117	-.01160	.51443	.10568	.17759	-.23276	.10568
9	.26480	.29008	-.20799	.23831	.49566	.00000	-.00001	-.49566
10	.44068	.13166	-.30055	-.24652	.30742	-.17759	.23276	.30742
11	.15403	-.11266	-.33259	-.02386	-.33097	.69579	.02987	-.33097
12	.12366	-.25845	.21662	.26408	.12462	.16302	.73814	.12462

Levels 9, 10, 11 and 12, non used in our study are not mentioned in this table

Table 4

A	<p><math>R = 0,17759</math></p>	<p><math>R = 0,11638</math></p>
B	<p><math>k = \sqrt{912} = 0,08333\dots</math></p>	<p><math>k = \sqrt{1420} = 0,0427949</math></p>
C	<p><math>k = \sqrt{72} = 0,5723502</math></p>	<p><math>k = \sqrt{105} = 0,0915869</math></p>
D	<p><math>R = 0,122678</math></p>	<p><math>k = 0,0819924</math></p>
E	<p><math>k = \sqrt{720} = 0,236067</math></p>	<p><math>k = 0,188982</math></p>

Table 5

The eigenvector of each pair (table 5) is orthogonal to its associate and also to the others eigenvectors of the matrix.

### $\pi$ ELECTRON DENSITY (H.M.O. METHOD)

We consider an alternant hydrocarbon, containing  $2n$  carbon atoms. We recall that  $x_{i\mu}^2$  and  $x_{i\mu}^2$ , diagonal elements of  $P_{\mu}^{\prime}$  resp.  $P_{\mu}^{\prime\prime}$  are depending of the choice of pair of eigenvectors (see chap. III and IV),  $x_{i\mu}^2 + x_{i\mu}^2$ , diagonal element of  $P_{\mu}$  is independent of this choice. Four ground state and closed shell molecules, each of the lower levels contains 2 electrons; the  $\pi$  electron density H.M.O. Method of atom  $i$  is given by (18)

$$(18) \quad q_i = 2 \sum_{\mu=1}^n x_{i\mu}^2 = 1$$

If the level  $y_{\mu}$  with  $\mu < n$  is doubly degenerate, the sum  $x_{i\mu}^2 + x_{i\mu}^2$  is independent of the choice of pair of eigenvectors; But if the highest occupied level  $y_{\mu}$  is doubly degenerate (see our example  $y = 9$ ), the relation (18) is valid only, if each of the two levels contains 1 electron.

Numerical example: Atom 2 (see table 4)

The sum of the electron densities of levels 1 to 5 is  
 $2(0,27769^2 + 0,37111^2 + 0,11597^2 + 0,24023^2 + 0,20634^2) = 0,6571396$   
 $2 x_{2\mu}^2 = 2 \cdot 0,5182^2 = 0,5370624$   
 $2 x_{2\mu}^2 = 2 \cdot 0,27263^2 = 0,1486542$   
 $x_{2\mu}^2 + x_{2\mu}^2 = 0,3428583$   
 $0,6571396 + 0,3428583 = 1$

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